

Synthesis and Photophysics of Phenylene Based Triplet Donor–Acceptor Dyads: *ortho* vs. *para* Positional Effect on Intramolecular Triplet Energy Transfer

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ABSTRACT

Two phenylene based geometrical/isomeric triplet *ortho*- and *para*-dyads (*o*-3 and *p*-3, respectively) were synthesized and fully characterized using advanced photophysical tools and computations. In dyad *o*-3, the through-space donor-acceptor interactions led to simultaneous triplet energy transfer and charge transfer with identical kinetics. On the other hand, in the dyad *p*-3, it was found that the phenylene spacer favors a fast triplet energy delocalization over the charge transfer process. Furthermore, analysis of the results from the present investigation indicates that the deactivation of the photo-excited species (*o*-3)* occurs through both the intrinsic channel *viz.* $S_0 \leftarrow S_1$ and charge recombination. In the case of dyad *p*-3, the results indicate that the primary deactivation pathway is self-quenching or triplet-triplet annihilation involving the acceptor unit(s).

1. INTRODUCTION

The efficiency of light-induced triplet energy transfer (TEnT) in organic chromophores is often dictated by the extent of overlap of the interacting chromophores' orbitals (wavefunctions). [1, 2, 3, 4] In this regard, it has been established that molecular interaction(s) in triplet donor/acceptor dyads systems can ultimately influence the resulting efficiency or performance of organic photo-materials and devices that are constructed from these molecular systems. [5, 6] Hence, in order to maximize light-harvesting and triplet energy flow dynamics and/or kinetics within these photo-materials/devices, it is essential to engineer donor-acceptor chromophoric systems that exhibit complementary structural features and/or optoelectronic bandgap. Recent proposals to achieve efficient donor→acceptor TEnT include the use of i) donor and acceptor chromophores with closely related structures, or ii) molecular

templates that link the interacting donor/acceptor chromophores. [7] Capitalizing on the latter approach, chemists have been able to synthesize many triplet bichromophoric systems, which were tailored for applications such as triplet-triplet annihilation based photon upconversion, [8, 9] photocatalysis, [10, 11] bioimaging, [12] and photodynamic therapy. [13, 14, 15, 16, 17, 18, 19]

While the tethering of the donor-acceptor chromophores onto molecular templates is necessary to achieving the ideal photophysical process, this approach may not be sufficient to realize faster TEnT kinetics. In such a design, the through-space or through-bond donor→acceptor triplet energy flow is certainly dependent on the position and orientation of the acceptor chromophores with respect to the triplet energy donor. [7, 20] To elucidate the positional and/or orientational effects on the photo-excited behavior and kinetic of triplet energy flow in organic bichromophores or dyads, we have designed *ortho*-

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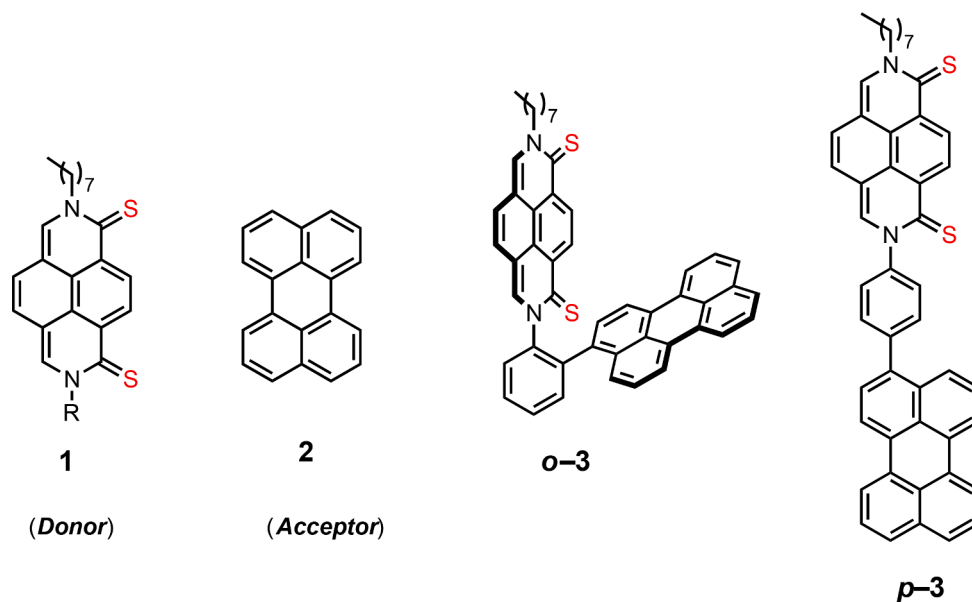


Fig. 1. Chemical structures of QDN (1), perylene (2), dyads *o*-3 (ortho-dyad 3) and *p*-3 (para-dyad 3). R = phenyl or $n\text{-C}_8\text{H}_{17}$.

and *para*-phenylene dyads using recently reported triplet energy donor and acceptor molecular systems from our group. [21, 22] Previously, we reported that a quinoidal naphthothioamide (QDN) triplet sensitized (1) can be used to sensitize perylene (2) leading to perylene delayed fluorescence via triplet-triplet annihilation. From this prior investigation, it was hypothesized that due to their identical structural features (*viz.* planar backbone), chromophores 1 and 2 would form $\pi\text{-}\pi$ supramolecular solid-solutions. After failing to prepare ideal co-crystalline solid materials using mixtures of chromophores 1 and 2, it became clear to us that the best strategy to create the desired materials is to tether sensitizer 1 and perylene 2 onto a rigid template such as phenylene or triptycene. Following this proposition, we reported recently molecular dyads, where the two chromophores were linked using i) a phenylene spacer which allow *meta*-interaction between the donor and acceptor moieties [23] and ii) a triptycene linker for through-space interaction. [24]

While the *meta*-dyad exhibited similar photophysical features as the parent sensitizer 1, the donor-acceptor interaction in the phenylene based dyad produced essentially a charge-transfer (CT) state upon excitation. Herein, we wish to report complementary scenarios with dyads *o*-3 and *p*-3 (Figure 1), where it was hypothesized that the *ortho*- and *para*-positional effects from the 1/2 interaction could modulate the extent of ground state coupling and excited state triplet energy flow or CT dynamic.

2. METHODS

2.1. Experimental Section

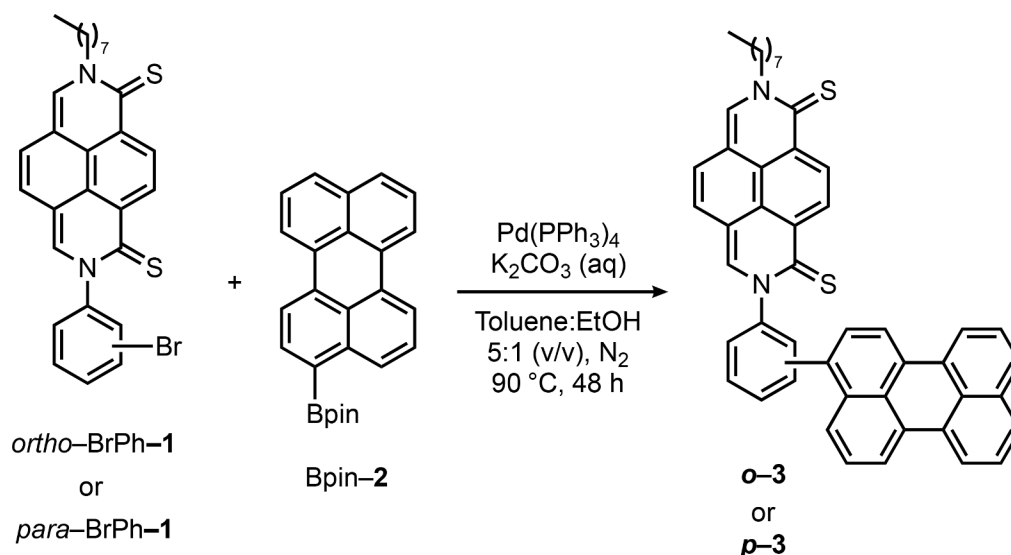
The synthetic procedures for *ortho/para*-BrPh-1 and Bpin-2 and their corresponding precursors are reported in the Supplementary Information (Schemes S1–S8). All spectroscopy measurements were performed using spectroscopy-grade solvents. All NMR characterizations were carried out on a Bruker 300 MHz spectrometer at 298 K. High-resolution mass spectrum data were recorded on a Bruker microTOF II or Shimadzu IT-TOF spectrometers in positive (ESI+) ion mode. UV-vis absorption spectra were recorded on an Ocean Optics spectrometer (DH-MINI UV-vis-NIR light source and QE-Pro detector). Emission spectra were recorded on an Edinburgh Instrument FLS980 spectrometer. Time-resolved pump-probe spectroscopy was performed using an amplified Ti:sapphire laser system (Spectra Physics Spitfire) equipped with an optical parametric amplifier (OPA, Light Conversion, TOPAS). This

system produces 130 fs pulses at 5 kHz centered at 800 nm. 95% of the output from the amplifier is directed to the OPA to generate tunable pump pulses in the visible and near-infrared spectral regions. For operation with 130 fs temporal resolution, the pump pulse and the remaining 5% of output from the amplifier are directed to a transient absorption spectrometer (Helios from Ultrafast Systems), where the 5% output is used to generate a continuum probe pulse extending from 450 nm to 1, 400 nm by focusing into a thin sapphire window. The pump pulse is chopped at half the repetition rate to measure a difference spectrum for the transient absorption measurement. The incident pump pulse for these experiments at 470 nm or 520 nm had energy on the sample of 400 nJ per pulse, focused to a 200- μm -diameter spot. The transmitted probe light was collected and fiber optically coupled to a spectrograph that used a visible (Si) array detector. Data were collected for continuum wavelengths from 450 nm to 750 nm as a function of delay track position for the continuum probe relative to the undelayed pump pulse. The temporal chirp of the data was experimentally determined and corrected before analysis. For longer time scale processes, the probe light comes from a continuum light source (EOS from Ultrafast Systems). In this case, the system operates at 1 kHz and has a time resolution of 200ps/point. Decay times of several hundred microseconds can be measured.

2.2. Calculation Details

The geometries of all adducts were optimized with help of hybrid correlation-exchange parameter-free functional of Perdew-Burke-Erzerhof (PBE0). [25, 26] All atoms were described by correlation-consistent basis sets of triple- ζ quality (cc-pVTZ). [27] In all cases, no symmetry restrictions were applied. All these calculations were performed using the Firefly program (version 8.1.0, A. A. Granovsky, Firefly <http://classic.chem.msu.su/gran/firefly/index.html>)

Using PBE0/cc-pVTZ-optimized geometries, a set of theoretical descriptors/indexes of aromaticity was calculated. This set includes Nuclear Independent Chemical Shift (NICS, introduced by von Rague Schleyer *et al.* [28, 29]), and calculations of NICS values were performed using Gauge Independent Atomic Orbitals (GIAO) approach with help of Gaussian09 program [30] at the PBE0/cc-pVTZ level of theory. Supplementary Information (Tables S1-S4) The set of descriptors was augmented by detailed consideration of magnetic induced ring current on target systems using Anisotropy of the Induced Current Density



Scheme 1. Synthesis of phenylene-based donor-acceptor dyad *o*-3 and *p*-3. The synthetic procedures of precursors/starting reagents are presented in the Supplementary Information Schemes S1–S8.

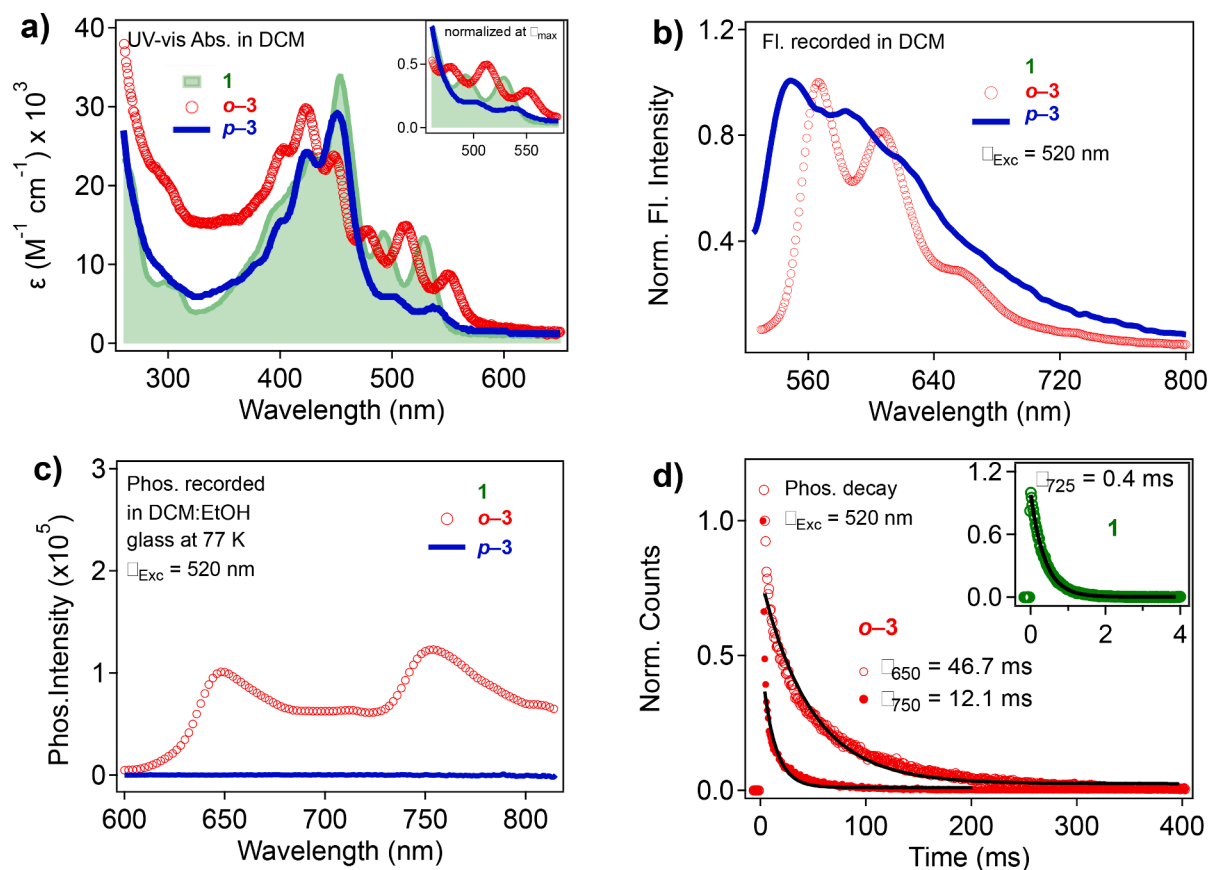


Fig. 2. (a) UV-vis absorption spectra of 1 (green), *o*-3 (red) and *p*-3 (blue) in DCM; (b) Normalized steady state emission profiles of 1 (green), *o*-3 (red) and *p*-3 (blue) in DCM; (c) Time-gated (90 μs delay after pulse) phosphorescence of 1 (green), *o*-3 (red) and *p*-3 (blue) in 50:50 (v/v) DCM:EtOH glass at 77 K; (d) Phosphorescence decay kinetics for 1 and *o*-3 in 50:50 (v/v) DCM:EtOH glass at 77 K at 1 Hz lamp repetition rate: samples O.D. = 0.15 ($\lambda_{\text{Exc}} = 520 \text{ nm}$).

(ACID) approach. [31] The applied magnetic field is perpendicular to the five-membered ring. To obtain induced current vectors and plot map, ACID 2.0.0 program uses the current density tensors, calculated by Continuous Set of Gauge Transformations (CSGT) method implemented in Gaussian 09 package. [32] Supplementary Information (Fig. S22)

3. SYNTHESIS

Synthesis of dyad *o*-3: *ortho*-BrPh-1 (130 mg, 1 equiv., 0.24 mmol) and Bpin-2 (138 mg, 1.5 equiv., 0.36 mmol) were added to a 50 mL toluene and 10 mL EtOH solution. Then, an aqueous solution of K_2CO_3 (3.8 mL (0.5M solution), 8 equiv., 1.94 mmol) was added to the reaction

Table 1
Optoelectronic parameters of QDN 1 and dyads *o*-3, *p*-3.

Comp.	λ_{Abs}^{max} (nm)	ϵ ($M^{-1}cm^{-1}$) ^a	λ_F^{max} (nm) ^b	λ_P^{max} (nm) ^b	τ_P (ms) ^{b,c}	ϕ_F (%)	ΔE_{L-H} (eV) ^d
1	454	34,070	552	725	0.4	0.20	3.18 ^e
<i>o</i>-3	422	29,814	566	755	46.7 & 12.1	0.67	3.07
<i>p</i>-3	450	29,184	560	n.d.	n.d.	0.60	3.18

^a ^b ^c ^d ^e UV-vis and FL were recorded in DCM; Phosphorescence was recorded in 50:50 (v/v) DCM:EtOH glass at 77 K. ^b $\lambda_{exc} = 520$ nm; ^cPhosphorescence decay kinetics. ^d $\Delta E_{L-H} = E_{LUMO} - E_{HOMO}$; Calculations were performed using geometries optimized at Calculations were performed at the PBE0/cc-pVTZ level of theory. ^eThe ΔE_{L-H} for QDN 1 was taken from Ref 24.

^c Phosphorescence decay kinetics. ^d $\Delta E_{L-H} = E_{LUMO} - E_{HOMO}$; Calculations were performed using geometries optimized at Calculations were performed at the PBE0/cc-pVTZ level of theory. ^e The ΔE_{L-H} for QDN 1 was taken from Ref 24.

flask. This mixture was degassed by bubbling N₂ through the solution for 30 min. Next, Pd(PPh₃)₄ (30 mg, 11 mol%, 0.03 mmol) was added to the reaction flask. The new mixture was purged with a stream of N₂ and the solution was heated at 90°C for 48 h under constant stirring. After, the reaction solution was cooled to room temperature and quenched with 100 mL of brine. The organic phase was extracted with DCM. The combined organic fraction was dried over anhydrous Na₂SO₄ and concentrated in vacuo under reduced pressure using rotavap. The crude compound was purified by flash chromatography over silica gel: solvent system (30% DCM-70% Hexanes). The expected compound (*o*-3) was obtained as a red solid. Yield: 33 mg, 17%. **Scheme 1**

¹H NMR (300 MHz, CDCl₃) δ 9.35-9.33 (d, *J* = 6 Hz, 2H), 8.90-8.88 (d, *J* = 6 Hz, 2H), 8.50-8.46 (d, *J* = 12 Hz, 2H), 7.87-7.80 (m, 2H), 7.65-

7.63 (m, 2H), 7.52-7.49 (m, 5H), 7.00-6.98 (m, 6H), 4.80-4.77 (t, *J* = 9 Hz, 2H), 2.02-1.98 (m, 2H), 1.38-1.29 (m, 10H), 0.90-0.85 (t, *J* = 15 Hz, 3H).

(The ¹H NMR spectrum can be found in Figures S15).

HRMS (ESI-TOF) *m/z* [*M* + H]⁺ calcd for C₄₈H₃₉N₂S₂⁺ 707.2555, found 707.2549.

Synthesis of dyad *p*-3: Dyad *p*-3 was synthesized according to the procedure we developed for *o*-3 starting with *para*-BrPh-1 (150 mg, 1 equiv., 0.28 mmol) and Bpin-2 (160 mg, 1.5 equiv., 0.42 mmol). *p*-3 was isolated as a red solid. Yield: 48 mg, 23%.

¹H NMR (300 MHz, CDCl₃) δ 9.29-9.20 (m, 2H), 8.30-8.27 (m, 4H), 7.97-7.85 (m, 1H), 7.87-7.85 (m, 1H), 7.77-7.71 (m, 4H), 7.60-7.50 (m, 6H), 7.19-7.17 (m, 2H), 7.10 (s, 1H), 4.85-4.81 (t, *J* = 12 Hz, 2H), 2.08-2.04 (m, 2H), 1.39-1.31 (m, 10H), 0.92-0.87 (t, *J* = 15 Hz, 3H).

(The ¹H NMR spectrum can be found in Figures S16.)

HRMS (ESI-TOF) *m/z* [*M* + H]⁺ calcd for C₄₈H₃₉N₂S₂⁺ 707.2555, found 707.2549.

4. RESULTS & DISCUSSION

4.1. UV-vis Absorption and Emission Spectroscopy

As shown in **Figure 2a**, the UV-vis absorption profiles of *o*-3 and *p*-3 exhibit similar and/or combined transitions and vibronic features from both the sensitizer 1 and the acceptor 2 (Supporting Information, Fig. S17). Nevertheless, a distinctive feature in the absorption profiles of these dyads is the seemingly reduced molar absorption at $\lambda > 470$ nm, especially in the case of *p*-3 compared to the profiles of 1 and *o*-3 (**Figure 2a**: inset). It was hypothesized that this reduction in the absorption profile of *p*-3 is the result of through-bond electronic coupling between QDN 1 and the perylene unit 2. Certainly, the *para*-interaction

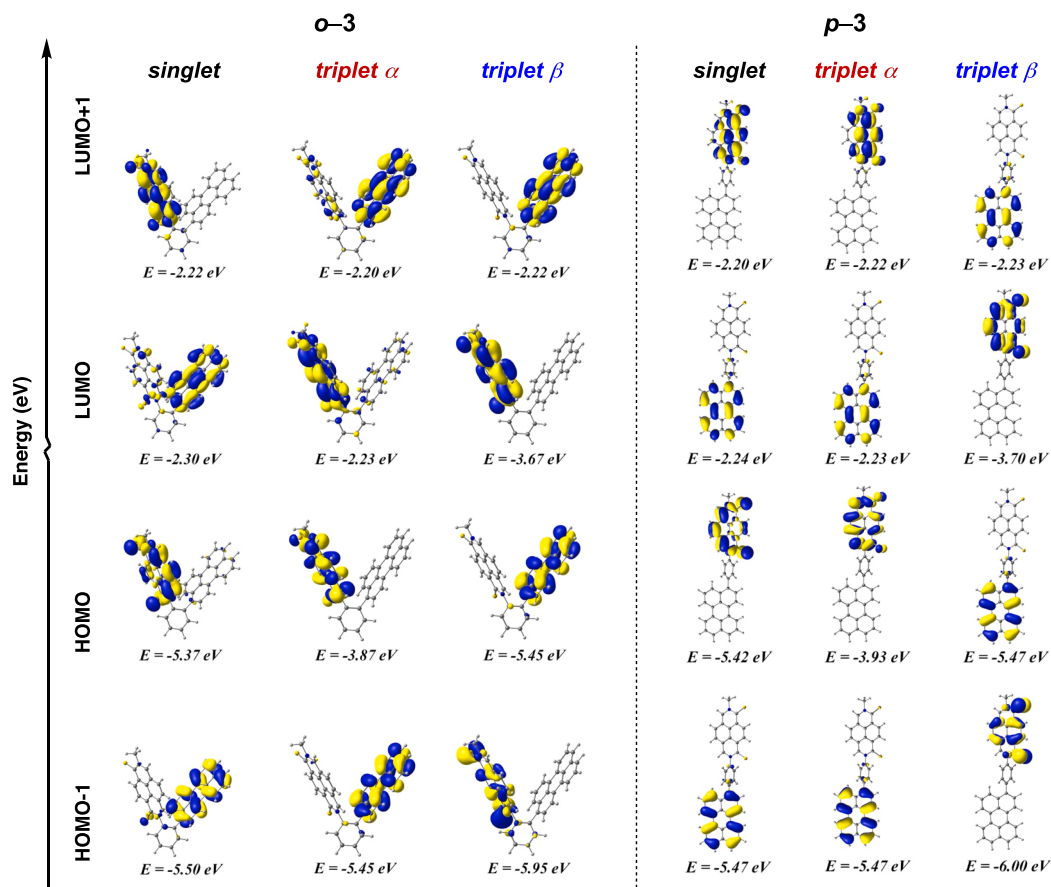


Fig. 3. Frontier molecular orbitals and energies of *o*-3 and *p*-3 dyads. Calculations were performed (for R = CH₃) at PBE0/cc-pVTZ level of theory.

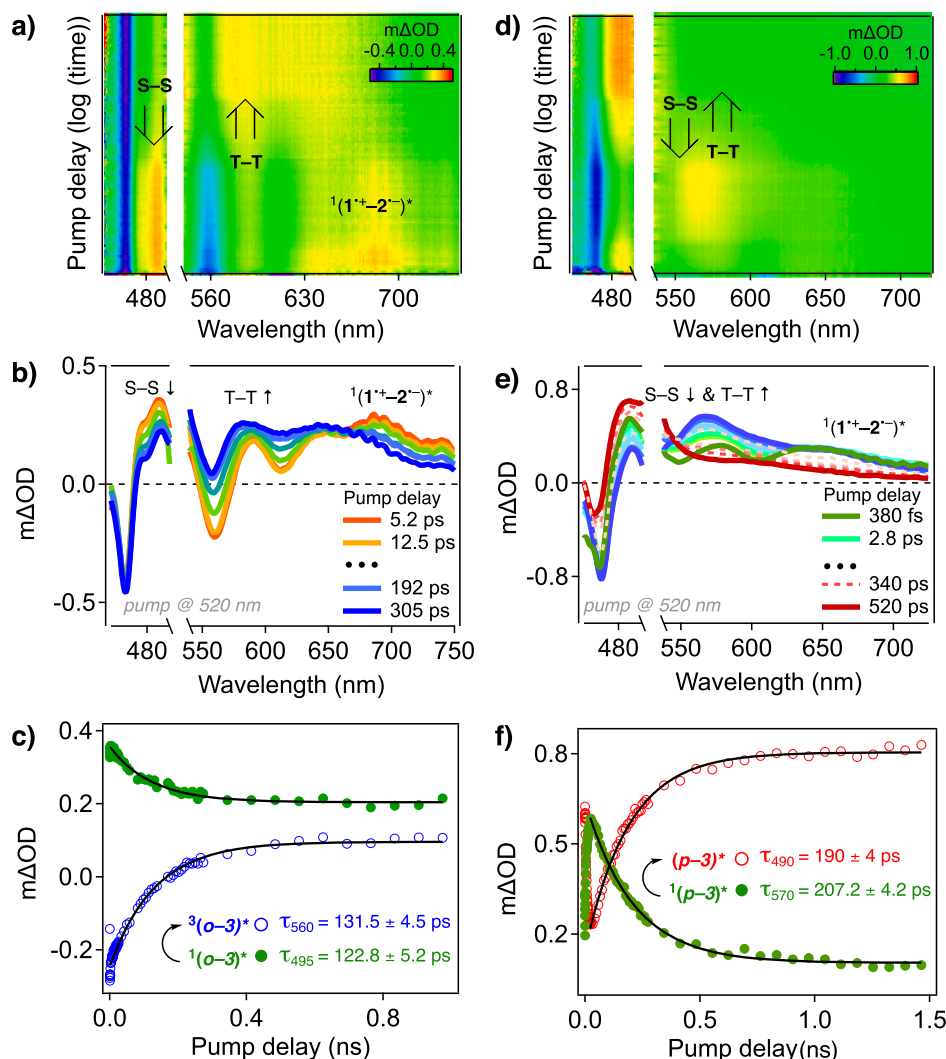


Fig. 4. (a & b) Femtosecond transient absorption map/spectra for *o*-3 and the corresponding kinetic traces (c); (d & e) Femtosecond transient absorption map/spectra for *p*-3 and the corresponding kinetic traces (f). Note: Samples for *o*-3 and *p*-3 were prepared in oxygen-free THF with optical density O.D. = 0.5 at $\lambda_{\text{Exc}} = 520$ nm; laser power = 0.4 $\mu\text{J}/\text{pulse}$ with 2.5 kHz repetition rate.

between QDN **1** and perylene unit **2** could also enhance π -conjugation throughout the whole molecular backbone; hence, the reduced absorption at $\lambda > 470$ nm is suggestive of “enhanced aromaticity” of the quinoidal ring of chromophore **1**, as elaborated in our previous publication. [22] For dyad *o*-3, the absorption profile resembles the one of the parent chromophore **1**; but, the appearance of a new transition at ca. $\lambda = 480$ nm is due energy shifts of the low energy transitions or an intramolecular ground state charge transfer process owing to the close proximity of the interacting chromophores (**1** and **2**). In this picture, we expected that the intra-dyad interaction 1/2 in the ground state could impact the excited state energy flow dynamics and the associated kinetics. To this end, we first recorded the steady-state emission of *o*-3 and *p*-3 (with $\lambda_{\text{Exc}} = 520$ nm) and compared the emission profiles to the one of **1**. Observation of the emission bands (Figure 2b) of the two dyads indicates that the perylene moiety exerts little effect on the singlet state behavior of the bichromophoric systems. Except the blue-shifted patterns, the fluorescence emission bands of the two dyads show similar vibronic features as seen in the profile of parent donor **1**. [22] Furthermore, the broad fluorescence emission band of dyad *p*-3 indicates the presence of an additional species, presumably from the $^1(\text{CT})^*$ state.

Interestingly, significant differences were found in the phosphorescence emissions of the two dyads (Figure 2c and 2d). While the emission

band of *o*-3 is blue-shifted with respect to the one of **1**, the triplet state of *p*-3 seems quenched, suggesting the following hypotheses: either i) the $(p-3)^*$ is trapped in the $(\text{CT})^*$ state (lower in energy than the lowest triplet state), or ii) triplet-triplet annihilation at the perylene moiety is contributing to the quenching of phosphorescence emission. In the two cases, one might expect an additional emission that overlaps with the steady-state radiative decay (Figure 2b). Per this hypothesis, it was concluded that the broad fluorescence emission band from *p*-3 is suggestive of two overlapped bands: one from $^1(p-3)^*$ and the other from either the $(\text{CT})^*$ or a delayed emission (*vide supra*). For dyad *o*-3, a closer look at the phosphorescence profile (Figure 2c and 2d) reveals various features that can be attributed to the convolution of emission bands from both $^3(1)^*$ and a triplet charge transfer species $^3(1^{*+}-2^{*-})^*$. With this assumption, we fitted the decay trace of the phosphorescence emission at wavelengths 650 and 750 nm using a mono-exponential function. And the fitting results produced two different lifetime values: $\tau_{650} = 46.7$ ms and $\tau_{750} = 12.1$ ms. The smaller lifetime was ascribed to the intrinsic lifetime from the triplet species of the sensitizer **1** within the dyad and the longest lifetime was attributed to the $^3(1^{*+}-2^{*-})^*$ species. Comparison of the opto-electronic parameters for dyads *o*-3 and *p*-3 are summarized in Table 1, where one can see that the increase in Φ_{F} value for the two dyads is suggestive of additional radiative decay, presumably from the $(\text{CT})^*$ state(s).

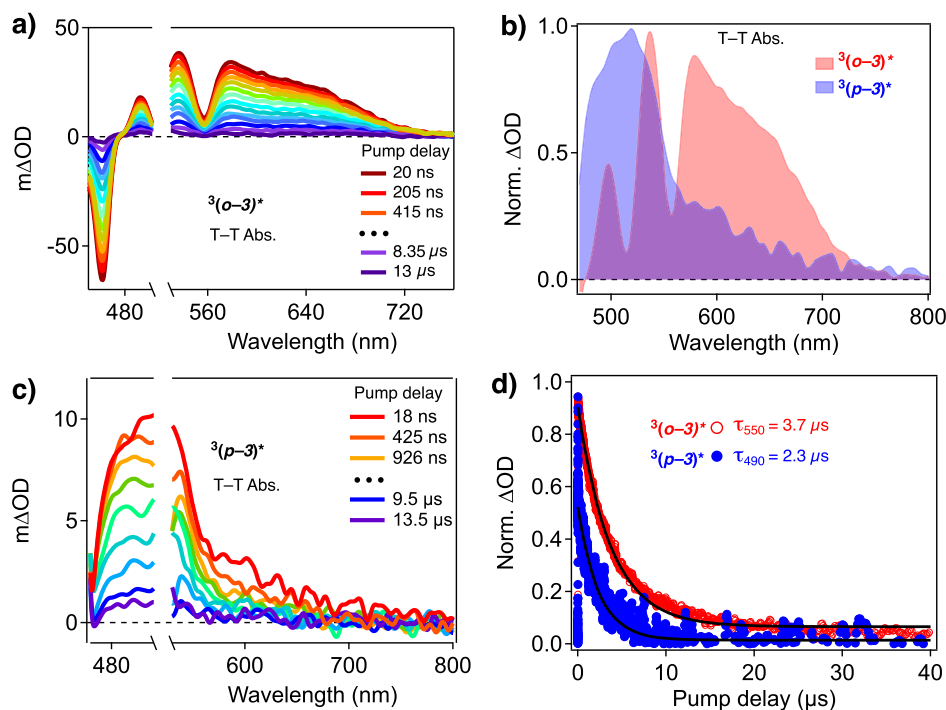


Fig. 5. Nanosecond transient absorption spectra for *o*-3 (a) and *p*-3 (c); (b) comparison of the transient spectra at time 18-to-20 ns after the laser pulse; (d) the corresponding normalized decay traces monitored at 550 nm (for *o*-3) and 490 nm (for *p*-3). Note: Samples for *o*-3 and *p*-3 were prepared in oxygen-free THF with optical density O.D. = 0.5 at $\lambda_{\text{Exc}} = 520$ nm; laser power = 1 $\mu\text{J}/\text{pulse}$ with 1 kHz repetition rate.

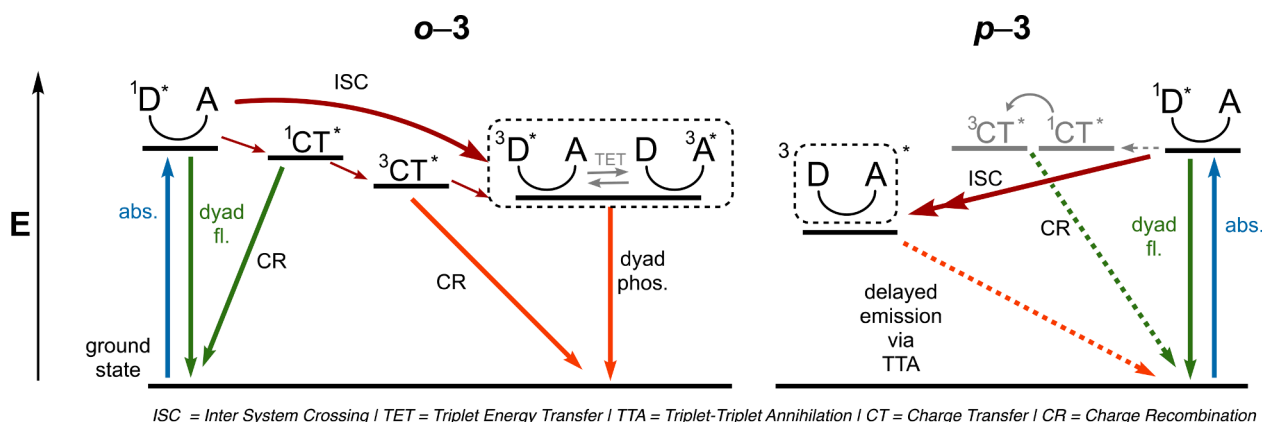


Fig. 6. Photophysical pathway of dyads with the proposed deactivation channels upon photo-excitation.: Note: The energy levels of the CT states were positioned hypothetically with respect to the lowest excited states.

4.2. Computational Investigations

Next, computational investigations were performed to decipher the extent of electron delocalization across the two dyads. The HOMO and LUMO molecular orbitals of *o*-3 and *p*-3 are presented in Figure 3. In the HOMO with singlet multiplicity, the electron density is essentially localized on the QDN moiety only, whereas in the LUMO the electron density is delocalized across the donor and acceptor moieties, indicating probable charge transfer dynamic in the upper-level electronic states. However, between the dyads, the degree of electronic communication between the donor and the acceptor moieties is greater in the case of *o*-3 than *p*-3. Thus, it can be concluded that the orientation and position of the individual chromophores in the dyad framework is pivotal to tuning the photo-excited state intra-dyad interactions or the expected triplet energy flow.

It is worth to mention that the nature of the triplet state, as described

in Figure 3, is determined by considering the presence/contribution of the perylene acceptor unit in the molecular skeleton of dyads *o*-3 and *p*-3. Also, the results from the calculations revealed that the β -MO is exclusively localized on the acceptor perylene unit (for both systems, *o*-3 and *p*-3). This finding is in full agreement with the electronic structure of the neutral systems, specifically with the topology of their LUMOs and also makes a clear bridge to the nature of the corresponding first singlet excited states.

4.3. Transient Absorption Spectroscopy

To have a better understanding of the photo-dynamics in the two dyads, we further investigated their corresponding excited state transients and the associated kinetics. Because of the observed dichotomy in the absorption and emission profiles, it was expected that the transient absorptions would also reveal significant differences in photo-behaviors.

In the femtosecond transient absorption spectra for both dyads (Figure 4), the negative absorption at ca. 470 nm is undoubtedly the ground state depletions, which overlap with the UV-vis absorption spectra. For *o*-3, the positive absorption at ca. 495 nm was ascribed to the singlet transient $^1(\mathbf{o-3})^*$, which is quasi-isoenergetic with $^1(\mathbf{1}^{*+}-\mathbf{2}^{*-})^*$ at ca. 700 nm with a decay kinetic of $\tau_{495/700} = 122.8$ ps. Expectedly, during the decay of the singlet species, a new broad band concomitantly appears at 560-650 nm. The band center at ca. 560 nm has always been assigned to the intrinsic triplet transient $^3(\mathbf{1})^*$; [24] however, with the degree of orientation and the short distance separating the donor and acceptor chromophores in this dyad, it can be ascertained that the additional band at ca. 650 nm comes from the formation of the $^3(\mathbf{1}^{*+}-\mathbf{2}^{*-})^*$ transient (Figures 4a, b). From the above observations and the earlier hypotheses of the excited state interactions with regard to the proximity of the QDN to the perylene moiety in dyad *o*-3, we can conclude that the similarity between the temporal decay of the singlet transient and the rise of the triplet transient is facilitated by the CT species $(\mathbf{1}^{*+}-\mathbf{2}^{*-})^*$.

Unlike *o*-3, the femtosecond transient spectra of dyad *p*-3 showed a combination of bands at 480, 560, and 650 nm (Figures 4d, e). However, it is easier to see that the earlier/fast kinetic (green spectrum at 380 fs after laser excitation) encompasses many events *viz.* pseudo-decay of the $^1(\mathbf{p-3})^*$ and $^1(\mathbf{1}^{*+}-\mathbf{2}^{*-})^*$ and rapid rise of the corresponding $^3(\mathbf{p-3})^*$ and $^3(\mathbf{1}^{*+}-\mathbf{2}^{*-})^*$. It was not easy to fully deconvolute these events, as these overlap with the instrument-response-time. At longer timescales, all of these transient species seem to co-exist with a decay time constant (at 570 nm) of 207 ps and a time constant of 190 ps for the rise of the band at ca. 490 nm (Figures 4d, e). With these complexities in the photo-excited state of dyad *p*-3, we hypothesized that although the triplet energy delocalization across the whole dyad is the preferred process/channel, the formation of the $^3(\mathbf{CT})^*$ species can still occur. But these species will eventually decay via charge recombination.

The nanosecond transient absorption spectra of the two dyads were recorded to monitor the decay of the triplet transients and $(\mathbf{CT})^*$ species. For *o*-3, since the $^3(\mathbf{o-3})^*$ transient and the $^3(\mathbf{1}^{*+}-\mathbf{2}^{*-})^*$ form within the same time constant from the singlet transients, it was concluded that the triplet spectrum evokes mixed radiative $S_0 \leftarrow T_1$ and charge recombination processes with a time constant of 3.7 μ s. The scenario is different with *p*-3, where the possibility of triplet energy flow ($\mathbf{1} \rightarrow \mathbf{2}$) can lead to the formation of the perylene triplet transient. In this case, we assigned the band centered at ca. 490 nm to perylene triplet-triplet absorption, which a time constant of 2.3 μ s (Figures 5c, d).

The overlapped triplet-triplet absorption spectra of the two dyads in Figure 5b does corroborate our hypothesis that *ortho*-geometry (in these donor-acceptor chromophoric systems) could lead to both triplet energy flow and charge transfer processes, whereas *para*-geometry favors not only triplet delocalization across the whole molecule, but also the rapid quenching of the triplet at the perylene moiety. The photophysical processes in the two dyads are summarized in Figure 6.

5. CONCLUSION

The positional effect on the photo-dynamics in the dyads under study revealed that donor-acceptor interactions can lead to both triplet energy flow and charge transfer processes when the chromophores interact through space in a co-facial manner. On the other hand, when these chromophores interact through the phenylene spacer, the triplet energy flow becomes the predominant process. In dyad *o*-3, the energy and charge transfer processes occur simultaneously; whereas in dyad *p*-3, the excited-state dynamic is dominated by triplet energy delocalization donor \leftrightarrow acceptor, which ultimately leads to triplet-triplet quenching at the perylene moiety. Importantly, the photophysical results are in agreement with the computational investigations. The present result provides complementary understanding of geometrical influences on the photophysics of molecular bichromophores, which could be employed for a number of light-harvesting and -modulation processes.

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ASSOCIATED CONTENT

Supporting Information: Details of the synthetic procedures for all precursors of dyads *o*-3 and *p*-3: ^1H and ^{13}C NMR spectra, additional UV-vis absorption spectra, emission spectra, IR spectra, and computational data. This material is available free of charge via the Internet at <http://elsevier.com>.

Declaration of Competing Interest

The authors declare no competing financial interests.

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Supplementary materials

Supplementary material associated with this article can be found, in the online version, at [doi:10.1016/j.jpap.2022.100112](https://doi.org/10.1016/j.jpap.2022.100112).

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